

ISOLATION AND STRUCTURE OF AKAGERINE: A NEW TYPE OF INDOLE ALKALOID

L. Angenot*, O. Dideberg⁺ and L. Dupont⁺

* Institut de Pharmacie, Université de Liège, rue Fusch, 5, B-4000 Liège, Belgium

+ Institut de Physique, Université de Liège au Sart-Tilman, B-4000 Liège, Belgium

(Received in UK 3 March 1975; accepted for publication 13 March 1975)

The roots of *Strychnos usambarensis* are used as arrow poison in Central Africa. Early investigations have shown that the active principles were divided into tertiary and quaternary alkaloids¹. We now record the isolation of an additional tertiary alkaloid: akagerine².

The tertiary crude bases were purified by a column chromatography over Al₂O₃ (activity II-III) and eluted with ether. 20- ml fractions were collected. After check TLC, the fractions 72-98 were combined and gave amount of the pure akagerine which crystallized in hexane, as plaquettes; m.p. 188°C (dec.); ORD (c:0,004-MeOH) [ϕ]+8100° at 280-284 nm; CD (MeOH) [θ]₂₆₅ = +13200.

The molecular formula has been established by mass spectrometry (high resolution) = m/e (relative abundance) 324 M⁺[C₂₀H₂₄N₂O₂] (51), 309(9), 306 [C₂₀H₂₂N₂O] (44), 277[C₁₉H₂₁N₂] (21), 263 (13), 241 (14), 223 (13), 214 (30), 213 [C₁₃H₁₃N₂O] (36), 198 [C₁₃H₁₂NO] (27), 186 (11), 185 [C₁₂H₁₃N₂] (100 = base peak), 184 (56), 183 (28), 180 (18), 172 (13), 171[C₁₁H₉NO] (80), 156 (11), 144 (8), 143 (10), 130 (6).

The U.V. spectrum [λ _{nm}^{MeOH} (log ϵ)] 227 (4.51), 276 (3.82), 283 (3.82), 293 (3.73)] indicates the presence of an indole (tetrahydro- β -carboline) system.

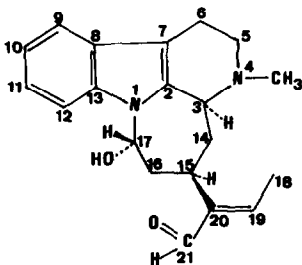
The I.R. spectrum [ν _{max}^{KBr}] 3450, 3050, 2930, 2850, 2700, 1680, 1628, 1445, 1370, 1350, 1305, 1295, 1280, 1200, 1110, 890 and 742 cm⁻¹], shows bands which can be assigned to unsaturated carbonyl (1680) and indole (742).

The N.M.R. spectrum (in CDCl₃) shows characteristic signals at δ 9.19 (aldehydic proton), δ 6.45 and δ 2 (ethylidene side chain), δ 2.33 (N-CH₃), δ 7.4-7 (four aromatic protons), δ 4.45 (proton exchangeable with D₂O).

Mild acetylation (Ac₂O-C₅H₅-RT) afforded the acetylated akagerine which showed characteristic absorption of OAc bands in the I.R. and N.M.R. spectra. The lack of proton exchangeable with D₂O is also obvious.

X-Ray crystallography of the base has confirmed the previous results but has mainly established the relative configuration of akagerine. Crystals of akagerine recrystallized from hexane are tetragonal, space group P 4₁2₁2 or P 4₃2₁2 with 8 molecules in an unit cell of dimensions a = b = 9.255 Å, c = 42.165 Å.

Akagerine is a tetracyclic indole alkaloid possessing a perhydroazepine ring coupled to tetrahydro- β -carboline, by an original N₁C₁₇ bond³. Indeed, it is the first time that such an alkaloid has been found. Moreover, a cis-relationship between H₃, H₁₅, OH₁₇ and the electrons lone pair of N₄ is detected by this X-Ray analysis⁴.

AKAGERINE $C_{20}H_{24}O_2N_2$

Examination of molecular stereomodels formed by condensation of N_N -methyl-tryptamine and a monoterpene unit derived from loganic acid, has shown that the 15α -configuration for $C_{15}H$ agrees with the biosynthetic hypothesis ^{5 to 8}. From these data, the complete absolute stereochemistry of akagerine could be depicted as that shown above.

Akagerine should be the precursor of more sophisticated alkaloids (bisindole and heptacyclic compounds) that would be obtained by reaction with tryptamine. These new products would be different from usambarensine ^{1,9}.

The results of this research are presented to Prof. A. Denoël for his 60th birthday.

Acknowledgements: We thank sincerely Drs N.G. Bisset, P.J. Hylands, as well MM. D. Carter and G. MacDonough (University of London) for their collaboration in this work.

References and complementary notes

1. L. Angenot : Diss. Abstr. Intern., 34, 11, 338 (1974).
2. The name akagerine was chosen because *Strychnos usambarensis* was collected in the National Park of Akagera (Rwanda), where the plant is abundant.
3. In this letter, the numbering system is that of J. Le Men and W.I. Taylor : *Experientia*, 21, 508-510 (1965).
4. For full details about cristallography, see: L. Dupont, O. Dideberg and L. Angenot : *Acta Cryst.* (to be published in 1975).
5. E. Wenkert and N.V. Bringit : *J. Amer. Chem. Soc.*, 81, 1474-1481 (1959).
6. A.R. Battersby in "The Alkaloids-Specialist Periodical Reports", The Chemical Society, London, 1, 31-47 (1971).
7. L. Angenot and A. Denoël : *Pl. medicin. et Phytothérapie*, 1, 284-292 (1973).
8. R. Guarnaccia, L. Botta and C.J. Coacia : *J. Amer. Chem. Soc.*, 96, 7079-7084 (1974).
9. O. Dideberg, L. Dupont and L. Angenot : *Acta Cryst.* (1975) under press.